

This article was downloaded by:

On: 22 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713454007>

### **A new cerebroside, Asperiamide A, from the marine fungus *Asperillus sp.***

Ming-An Ouyang<sup>a</sup>; Rong Liu<sup>a</sup>; Yueh-Hsiung Kuo<sup>b</sup>

<sup>a</sup> Department of Bio-engineering & Technology, Huaqiao University, Quanzhou, Fujian, China <sup>b</sup>

Department of Chemistry, Taiwan University, Taipei

**To cite this Article** Ouyang, Ming-An , Liu, Rong and Kuo, Yueh-Hsiung(2005) 'A new cerebroside, Asperiamide A, from the marine fungus *Asperillus sp.*', *Journal of Asian Natural Products Research*, 7: 5, 761 – 765

**To link to this Article:** DOI: 10.1080/1028602042000324853

**URL:** <http://dx.doi.org/10.1080/1028602042000324853>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## Note

# A new cerebroside, Asperiamide A, from the marine fungus *Asperillus sp.*

MING-AN OUYANG<sup>†\*</sup>, RONG LIU<sup>†</sup> and YUEH-HSIUNG KUO<sup>‡</sup>

<sup>†</sup>Department of Bio-engineering & Technology, Huaqiao University, Quanzhou, Fujian 362011, China

<sup>‡</sup>Department of Chemistry, Taiwan University, Taipei

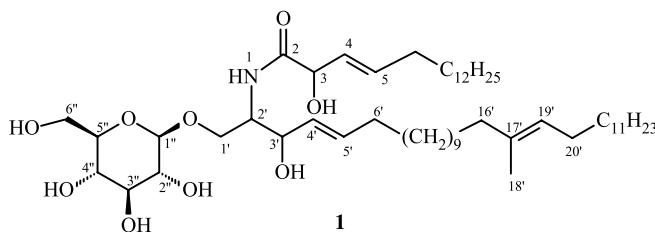
(Received 12 April 2004; revised 1 July 2004; in final form 27 August 2004)

A new cerebroside, asperiamide A (**1**), and adenosine were produced from a marine fungus, identified as *Asperillus sp.*, collected in the Mei Zhou Gulf, China. Two-dimensional NMR methods, FAB-MS, were used to establish the structure of the new compound.

**Keywords:** Marine fungus; *Asperillus*; Cerebroside

## 1. Introduction

Cerebrosides and ceramides [1] have been isolated from a number of marine organisms such as sea stars, sea anemones, gorgonians, sponges, tunicates, dinoflagellates, and green algae. Some cerebroside and ceramides exhibited cytotoxic, antitumor [2,3], immunostimulatory [4], antifungal [5], antiviral [6]. In the search for bioactive components, two water soluble constituents, asperiamide A and adenosine [7], were isolated from the marine fungus *Asperillus sp.*, the current report describes the structural elucidation of a new compound, asperiamide A (**1**) and a known one, adenosine.



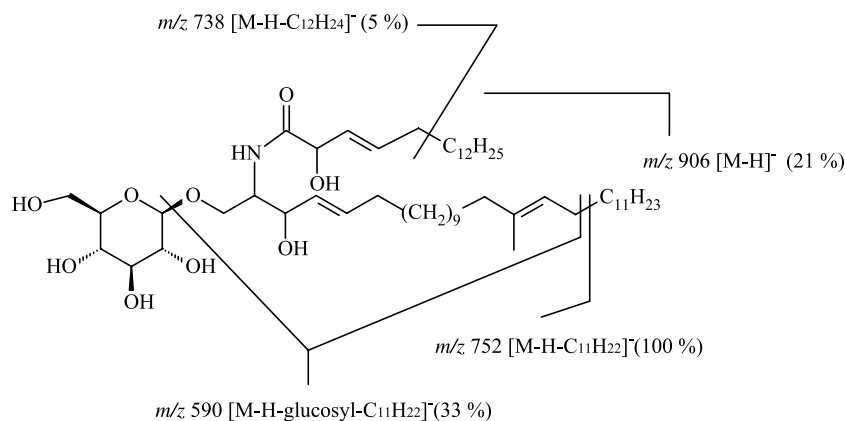
\*Corresponding author. Tel.: +86-595-22693976. Fax: +86-595-22693685. Email: maouyang@hqu.edu.cn

## 2. Results and discussion

Asperiamide A (**1**) was obtained as a white amorphous powder. Its molecular formula was determined as  $C_{54}H_{101}NO_9$  on the basis of HRFABMS. There were five degrees of unsaturation in the molecule according to the molecular formula. Compound **1** exhibited a molecular ion peak at  $m/z$  906  $[M-H]^-$  in the negative FABMS. IR data of **1** indicated that it contains  $-OH$ ,  $-NH$  ( $3368\text{ cm}^{-1}$ ),  $C-H$  ( $-CH_2-$ ,  $-CH_3$ ,  $2910$ ,  $2856\text{ cm}^{-1}$ ), carbonyl ( $1649\text{ cm}^{-1}$ ), double bonds ( $1636\text{ cm}^{-1}$ ), methyl ( $1470$ ,  $1380\text{ cm}^{-1}$ ),  $C-N$  and  $C-O$  ( $1078$ ,  $1036\text{ cm}^{-1}$ ), tri-substitute vinyl ( $968\text{ cm}^{-1}$ ). The  $^1H$  and  $^{13}C$  NMR spectra of **1** (table 1) were consistent with the presence of a secondary amide group ( $\delta_H$  7.83, 1H, d,  $J = 9.2\text{ Hz}$ ,  $\delta_C$  52.9,  $\delta_C$  172.1) [8], and a glucopyranoyl group (table 1). Furthermore, the negative FABMS of **1** showed a quasimolecular ion at  $m/z$  906  $[M-H]^-$ ; The 17', 19'-double bond was assigned on the basis of the typical fragment ion at  $m/z$  752 (100%) which was formed by elimination of undecene through McLafferty rearrangement (figure 1), and 4, 5-double bond was deduced by elimination of dodecene from the quasimolecular ion, and the fragment peak was at  $m/z$  738  $[M-H-C_{12}H_{24}]^-$ . In the  $^1H$  NMR spectrum, the protons between  $\delta_H$  4.66–3.20 were assigned to protons attached to the carbons bearing a heteroatom. Two aliphatic chains were

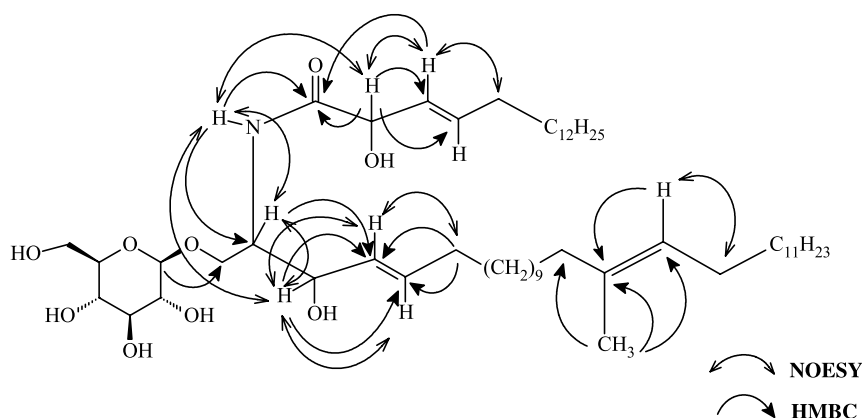
Table 1.  $^1H$  and  $^{13}C$  NMR data of asperiamide A (**1**) in DMSO- $d_6$ .

NO.	$^1H$	$^{13}C$	
1	7.83 (1H, d, $J = 9.2\text{ Hz}$ )		
2		172.1	C
3	4.79 (1H, d, $J = 7.2\text{ Hz}$ )	72.0	CH
4	5.74 (1H, dd, $J = 15.6, 7.2\text{ Hz}$ )	129.0	CH
5	6.05 (1H, dt, $J = 15.6, 6.2\text{ Hz}$ )	130.6	CH
6	1.94 (2H, overlap)	31.8	CH <sub>2</sub>
7–15	1.20–1.46	28.9–29.3	$-CH_2-$
16	1.25 (2H, m)	31.3	CH <sub>2</sub>
17	1.25 (2H, m)	22.4	CH <sub>2</sub>
18	0.84 (3H, t, $J = 7.1\text{ Hz}$ )	13.8	CH <sub>3</sub>
1'	3.95 (1H, dd, $J = 10.4, 3.2\text{ Hz}$ )	68.8	CH <sub>2</sub>
	4.35 (1H, dd, $J = 10.4, 6.0\text{ Hz}$ )		
2'	4.19 (1H, m)	52.9	CH
3'	4.32 (1H, d, $J = 7.2\text{ Hz}$ )	70.6	CH
4'	5.67 (1H, dd, $J = 16.0, 7.2\text{ Hz}$ )	130.9	CH
5'	5.92 (1H, dt, $J = 16.0, 7.2\text{ Hz}$ )	131.1	CH
6'	1.95 (2H, m)	32.2	CH <sub>2</sub>
7'–15'	1.20–1.46	28.9–29.3	$-CH_2-$
16'	1.88 (2H, overlap)	39.0	CH <sub>2</sub>
17'		134.7	C
18'	1.52 (3H, s)	15.7	CH <sub>3</sub>
19'	5.34 (1H, t, $J = 6.0\text{ Hz}$ )	123.6	CH
20'	1.93 (1H, m)	31.8	CH <sub>2</sub>
21'–28'	1.20–1.46	28.9–29.3	$-CH_2-$
29'	1.25 (2H, m)	31.3	CH <sub>2</sub>
30'	1.25 (2H, m)	22.4	CH <sub>2</sub>
31'	0.84 (3H, t, $J = 7.1\text{ Hz}$ )	13.8	CH <sub>3</sub>
1''	4.66 (1H, d, $J = 7.9\text{ Hz}$ )	103.5	CH
2''	3.42 (1H, t, $J = 7.9\text{ Hz}$ )	73.4	CH
3''	3.50 (1H, t, $J = 9.0\text{ Hz}$ )	76.9	CH
4''	3.54 (1H, m)	70.1	CH
5''	3.59 (1H, m)	76.6	CH
6''	3.89 (1H, dd, $J = 11.6, 3.2\text{ Hz}$ )	61.1	CH <sub>2</sub>
	4.06 (1H, dd, $J = 11.6, 4.0\text{ Hz}$ )		

Figure 1. The key negative FAB mass fragments of **1**.

suggested by the presence of signals for methylene groups at  $\delta_{\text{H}}$  1.20–1.46, and three methyl groups at  $\delta_{\text{H}}$  1.52 (s, 3H), and 0.84 (t, 6H). In the  $^{13}\text{C}$  NMR spectrum, an intense signal at  $\delta_{\text{C}}$  29–30 indicated methylene groups of aliphatic side-chains. Three double bond signals, including two disubstituted and one trisubstituted, were exhibited at  $\delta_{\text{C}}$  134.7 (C), 131.1 (CH), 130.9 (CH), 130.6 (CH), 129.0 (CH), and 123.6 (CH), and  $\delta_{\text{H}}$  6.05 (1H, dt,  $J = 15.6$ , 7.2 Hz, H-5), 5.92 (1H, dt,  $J = 16.0$ , 7.2 Hz, H-5'), 5.74 (1H, dd,  $J = 15.6$ , 6.2 Hz, H-4), 5.67 (1H, dd,  $J = 16.0$ , 7.2 Hz, H-4'), 5.34 (1H, t,  $J = 6.0$  Hz), these coupling constants ( $J = 15.6$  Hz and  $J = 16.0$  Hz) suggested that two disubstituted double bonds were *trans*. Compound **1** showed an anomeric proton  $\delta_{\text{H}}$  4.66 (1H, d,  $J = 7.9$  Hz) and anomeric carbon  $\delta_{\text{C}}$  103.5 suggesting  $\beta$ -configuration for the sugar which was detected as glucose by acid hydrolysis. The position of the sugar was confirmed by HMBC spectrum (figure 2).

The structural elucidation and assignments of complete proton and carbon signals were achieved by 2D NMR techniques and chemical methods. The carbonyl group position was assigned by the long-range HMBC correlations between H-3 at  $\delta$  4.79 and C-2 at  $\delta$  172.1, and between H-1 at  $\delta$  7.83 and C-2, between H-4 at  $\delta$  5.74 and C-2; the positions of three double bonds were also confirmed by the correlations of HMBC spectrum (see figure 2): between H-3 at  $\delta$  4.79 and C-4 at  $\delta$  129.0, C-5 at  $\delta$  130.6, between H-3' at  $\delta$  3.93 and C4' at  $\delta$  130.9, C-5' at  $\delta$  131.1, and

Figure 2. The Nosey and HMBC correations of **1**.

between the protons of methyl (H-18') at  $\delta$  1.52 and C-17' at  $\delta$  134.7, C-19' at  $\delta$  123.6. Based on the NOESY experiment (figure 2), the structure for **1** was further illustrated, the NOESY correlations were from H-1 to H-3, H-2', H-3', from H-4 to H-3, H-6, and from H-4' to H-3', H-6'. Thus the structure of asperiamide A was concluded to be **1**.

### 3. Experimental

#### 3.1 General experimental procedures

FAB-MS were taken on a VG Autospec 3000 system spectrometer.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were obtained with a Bruker AM-400 spectrometer and the solvent is  $\text{DMSO}-d_6$ . Optical rotations were taken on a JASCO-20C digital polarimeter, and the IR spectrum was recorded with a Perkin-Elmer 1750 FTIR spectrometer. Chromatographic stationary phase used RP-18 (40–60  $\mu\text{m}$ , Merck), silica gel (160–200 mesh), and Sephadex LH-20 (25–100  $\mu\text{m}$ , Pharmacia Fine Chemical Co., Ltd.). The following solvent systems were used:  $\text{CHCl}_3$ – $\text{MeOH}$ – $\text{H}_2\text{O}$  (80:20:2, and 70:30:5) and  $\text{MeOH}$ – $\text{H}_2\text{O}$  (0–100%) for the alkaloids. Spot of TLC was detected by spraying with 5%  $\text{H}_2\text{SO}_4$  followed by heating.

#### 3.2 Fungal isolation and culture

The fungal strain (culture MF-34#) was isolated from the sea water collected in Mei-Zhou Gulf, Fujian Province, China, in 2000 and identified as an *Asperillus sp.* by Prof. Chen Bi-E. The fungus was cultured (20 L) 25 days at 24°C in the medium: soluble starch (1%), yeast extract (0.2%),  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$  (0.05%),  $\text{KH}_2\text{PO}_4$  (0.1%), sea water (100%).

#### 3.3 Extraction and isolation

The mycelium and broth were separated by filtration. The mycelial mat was extracted with  $\text{MeOH}$ . The combined extract (5.1 g) was subjected to Dianon gel column and eluted with water and methanol. Evaporation of the methanol eluate yielded 1.3 g of a brown fraction. The fraction was chromatographed on silica gel to give four fractions. Each fraction was chromatographed on RP-18 gel column (solvent:  $\text{MeOH}$ – $\text{H}_2\text{O}$ , 10%–70%) and Si gel column using  $\text{CHCl}_3$ – $\text{MeOH}$ – $\text{H}_2\text{O}$  (80:20:2–70:30:5) as solvent to yield **1** (24 mg), **2** (87 mg).

#### 3.4 Acid hydrolysis of 1

Compound **1** (10 mg) was dissolved in 10 ml of a solution ( $\text{MeOH}$ : 1 M  $\text{H}_2\text{SO}_4$ —1:1) and refluxed for 6 h. The reaction mixture was neutralized with 2 N  $\text{NaOH}$  and extracted with  $\text{CHCl}_3$ , the aqueous layer was concentrated to dryness for identification. D-glucose was detected by HPTLC by comparison with authentic sample (solvent system:  $\text{CHCl}_3$ – $\text{MeOH}$ – $\text{H}_2\text{O}$ /7:3:0.5 (9 ml) +  $\text{HOAc}$  (1 ml)).

*Asperiamide A* (**1**) White amorphous powder,  $\text{C}_{54}\text{H}_{101}\text{NO}_9$ ,  $[\alpha]_D^{21}$  -16 (c 0.46,  $\text{MeOH}$ ); Negative FAB-MS  $m/z$  906  $[\text{M}-\text{H}]^-$  (21) 889 (3), 752 (100), 738 (5), and 590 (33); HRFABMS  $m/z$  906.7403 (calcd for  $\text{C}_{54}\text{H}_{101}\text{NO}_9$ )  $[\text{M}-\text{H}]^-$ ; IR  $\nu_{\text{max}}$ (liquid): 3368,

2910, 2856, 1649, 1636, 1540, 1470, 1380, 1078, 1036, 968  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR see table 1.

*Adenosine* Brown amorphous powder,  $\text{C}_{10}\text{H}_{13}\text{N}_5\text{O}_4$ , Negative FAB-MS  $m/z$  266  $[\text{M}-\text{H}]^-$ ;  $^1\text{H}$  NMR (DMSO)  $\delta_{\text{H}}$ : 8.34 (1H, s, Ad-H-8), 8.12 (1H, s, Ad-H-2), 7.34 (2H, br. s,  $\text{NH}_2$ ), 5.86 (1H, d,  $J = 4.9$  Hz), 4.59 (1H, t,  $J = 4.2$  Hz), 4.13 (1H, br s), 3.95 (1H, br. d,  $J = 2.0$  Hz), 3.65 (1H, m), 3.52 (1H, m); and  $^{13}\text{C}$  NMR  $\delta_{\text{C}}$ : 156.2 (C), 152.4 (CH), 149.1 (C), 139.9 (CH), 119.4 (C), 88.0 (CH), 85.9 (CH), 73.5 (C), 70.7 (CH), 61.7 ( $\text{CH}_2$ ).

### Acknowledgements

This research was supported by a Chinese National Science Fund (Research Project: 20272015). We thank the staff of the analytical group of Department of Chemistry, Taiwan University for measurements of NMR and FAB-MS spectra.

### References

- [1] R.X. Tan, J.H. Chen. *J. Nat. Prod. Rep.*, **20**, 509 (2003).
- [2] T. Natori, Y. Koezuka, T. Higa. *Tetrahedron Lett.*, **34**, 5591 (1993).
- [3] T. Natori, M. Morita, K. Akimoto, Y. Koezuka. *Tetrahedron*, **50**, 2771 (1994).
- [4] M. Seki, A. Kayo, K. Mori. *Tetrahedron Lett.*, **42**, 2357 (2001).
- [5] H.Y. Li, S. Matsunaga, N. Fusetani. *Tetrahedron*, **51**, 2273 (1995).
- [6] G.T. Carter, K.L. Rinehart. *J. Am. Chem. Soc.*, **100**, 7441 (1978).
- [7] E. Pretsch, P. Bühlmann, C. Affolter. *Structure Determination of Organic Compounds Tables of Spectra Data*, pp. 291–292, Springer-Verlag, Berlin Heidelberg (2000).
- [8] S.Y. Kim, Y. Choi, H. Huh, J. Kim, Y.C. Kim. *J. Nat. Prod.*, **60**, 274 (1997).